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<b>(54) Title:</b> PROCESS FOR PRODUCING 1-OCTENE		
<b>(57) Abstract</b>  Process for producing 1-octene which comprises the steps of: (i) reacting 1,3-butadiene with a compound containing an active hydrogen atom and having formula R-H, in the presence of a telomerization catalyst to form a 1-substituted-2,7-octadiene; (ii) subjecting the 1-substituted-2,7-octadiene formed in step (i) to hydrogenation in the presence of a hydrogenation catalyst to form a 1-substituted octane; and (iii) decomposing the 1-substituted octane formed in step (ii) in the presence of a suitable catalyst to form 1-octene.		

# + DESIGNATIONS OF "SU"

Any designation of "SU" has effect in the Russian Federation. It is not yet known whether any such designation has effect in other States of the former Soviet Union.

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### PROCESS FOR PRODUCING 1-OCTENE

The present invention relates to a process for producing 1-octene.

1-Octene is used in large amounts as starting material for the production of various industrial products such as, for example, surfactant intermediates, plasticizer alcohols, lubricants and in the production of polymeric materials. For many of the industrial applications it is desirable to have 1-octene in a form as pure as possible, especially in the substantial absence of its branched olefinic C<sub>8</sub>-isomers, such as 2-methyl-1-heptene, 2-methyl-2-heptene, 3-methyl heptenes, 2-ethyl-1-hexene, dimethyl hexenes and similar compounds.

The most widely used commercial processes to produce 1-octene comprise oligomerization of ethylene to produce a range of linear  $\alpha$ -olefins as major products. By suitable selection of catalyst materials and process conditions the selectivity towards 1-octene can be optimized. However, as illustrated in ALPHA OLEFINS APPLICATIONS HANDBOOK, G.R. Lappin, J.D. Sauer, MARCEL DEKKER, INC. New York, pages 44-49, such oligomerization processes also produce some branched olefinic C<sub>8</sub>-isomers, such as 2-ethyl-1-hexene. As such by-product isomers frequently have a boiling point (2-ethyl-1-

hexene having a boiling point of 120.0 °C) which is very close to the boiling point of 1-octene (121.3°C), these by-products will be recovered in the 1-octene fraction isolated from the oligomerization reaction product. In view of this small difference in boiling point, it is very difficult to remove such isomers from the main product 1-octene by means of simple distillation techniques. The said isomers could be removed from the 1-octene product by means of adsorption-desorption separation techniques, but these techniques would increase the costs of the 1-octene product substantially and, accordingly, are not used in commercial practice to produce 1-octene. Therefore, the 1-octene processes commercially used generally yield 1-octene having a purity of at the most about 97.5%, whereas representative amounts of branched olefinic C<sub>8</sub>-isomers are between 0.5 and 1.5 wt%.

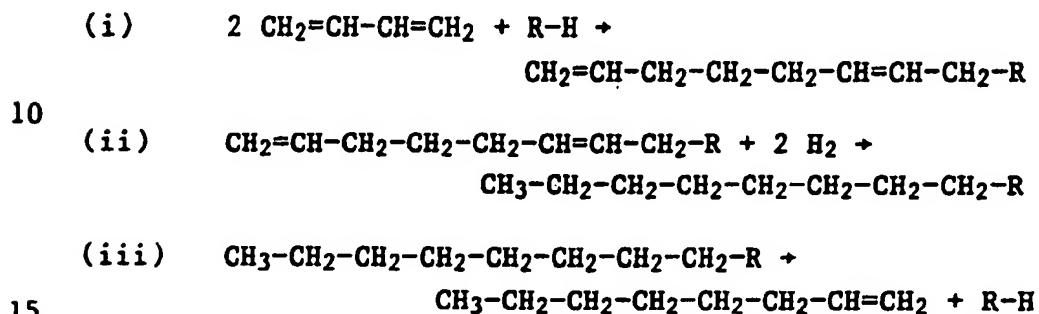
According to the present invention, by a careful selection of starting products and reaction steps, 1-octene containing substantially no branched olefinic C<sub>8</sub>-isomers can be obtained.

Accordingly, the present invention provides a process for producing 1-octene which comprises the steps of:

- (i) reacting 1,3-butadiene with a compound containing an active hydrogen atom and having formula R-H, in the presence of a telomerization catalyst to form a 1-substituted-2,7-octadiene of formula  $\text{CH}_2=\text{CH}-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}=\text{CH}-\text{CH}_2-\text{R}$ , in which R represents the residue of the compound containing an active hydrogen atom;
- (ii) subjecting the 1-substituted-2,7-octadiene formed in step (i) to hydrogenation in the presence of a hydrogenation catalyst to form a 1-substituted octane of

formula  $\text{CH}_3\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-R}$ ; and  
(iii) decomposing the 1-substituted octane formed in  
step (ii) in the presence of a suitable catalyst to form  
1-octene.

5           The above mentioned reaction steps are  
illustrated by the following equations:



The reaction in step (i) of the process of the present invention between an active hydrogen-containing compound R-H and 1,3-butadiene, also referred to as telomerization reaction, has been known per se for many years.

In step (i) of the present process any suitable telomerization process can be used which yields as primary product the compound of formula  $\text{CH}_2=\text{CH}-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}=\text{CH}-\text{CH}_2-\text{R}$  (1-substituted-2,7-octadiene). Suitable telomerization processes are for example described in US 3499042, US 3518315, US 3530187, GB 1178812, NL-A 6816008, GB 1248593, US 3670029, US 3670032, US 3769352, US 3887627, GB 1354507, DE-A 2040708, US 4142060, US 4219677, US 4146738, US 4196135 and EP-A 218100. The telomerization process may be carried out continuously or batch-wise.

The 1,3-butadiene used in step (i) may be employed as substantially pure butadiene or as a crude C<sub>4</sub>-hydrocarbons mixture. Such a crude C<sub>4</sub>-mixture contains in general, besides 1,3-butadiene other C<sub>4</sub>-hydrocarbons such as butenes and butanes. These other C<sub>4</sub>-hydrocarbons do not essentially influence conversion and selectivity of the 1,3-butadiene. Such a crude C<sub>4</sub>-hydrocarbon mixture can be produced as a by-product in the pyrolysis of naphtha, gas oil, LPG et cetera for the production of ethylene. As the selectivity and conversion towards the desired telomerization product, based on the 1,3-butadiene present in the crude C<sub>4</sub>-hydrocarbons mixture, are about the same as for pure 1,3-butadiene, it is advantageous to use the crude C<sub>4</sub>-mixture in step (i) of the present process, since doing so avoids the investments and costs associated with first extracting and purifying the 1,3-butadiene from the crude C<sub>4</sub>-mixture. The crude C<sub>4</sub>-mixture produced in the cracking of naphtha, gas oil or LPG usually has a 1,3-butadiene content of 20 to 70 % by weight. When the crude C<sub>4</sub>-mixture contains acetylenes, it is advantageous to selectively hydrogenate the crude C<sub>4</sub>-mixture in order to remove these acetylenes prior to its use in the present process.

The active hydrogen-containing compound R-H employed as co-reactant in step (i) of the process of the invention can be any compound having a reactive hydrogen atom. Suitable compounds include alcohols, hydroxy-aromatic compounds, carboxylic acids, ammonia, primary and secondary amines and silanols. The functional groups of these types of compounds contain active hydrogen atoms having a high reactivity towards 1,3-butadiene.

Exemplary of such alcohols are mono- or polyhydric alcohols containing primary OH-groups which alcohols can be linear or branched saturated compounds having up to 20 carbon atoms as well as unsaturated alcohols such as allyl alcohol; especially the mono- or polyhydric aliphatic alcohols having up to 8 carbon atoms, such as methanol, ethanol, propanol, butanol, ethylene glycol, propylene glycol, glycerol and the like. Secondary alcohols like isopropanol and the cycloaliphatic alcohols such as cyclopentanol and cyclohexanol may also be used advantageously.

Exemplary of hydroxy-aromatic compounds are aromatic compounds containing one or more rings such as phenol, benzyl alcohol, cresols, xylenols, naphtol, as well as polyhydric compounds such as resorcinol, hydroquinone and pyrocatechol. Also alkyl-, alkoxy- and/or halogen-substituted aromatic hydroxy compounds may be used, such as o-methoxyphenol, p-chlorophenol and the like.

Suitable examples of carboxylic acids include aliphatic carboxylic acids with up to about 20 carbon atoms. Preferred carboxylic acids are those having 1-6 carbon atoms such as e.g. acetic acid, propionic acid, butyric acid. Examples of suitable aromatic carboxylic acids include benzoic acid and toluene carboxylic acid. Representative examples of dibasic acids are adipic acid, the phthalic acids and the like.

Exemplary of suitable amine compounds are ammonia and primary and secondary amines. Suitable amine compounds include for example primary aliphatic amines, such as methylamine, ethylamine, butylamine, d decylamine and the like; primary aromatic amines, such

as aniline, toluidine, benzylamine and the like;  
secondary amines such as dimethylamine, diethylamine, N-  
methylaniline, dicyclohexylamine, methylhexylamine, and  
the like; as well as polyamines such as  
phenylenediamine, ethylenediamine; and heterocyclic  
5 amines, such as piperidine and the like.

In step (i) of the process of the invention, as  
the active hydrogen-containing compound preferably a  
primary aliphatic alcohol or aromatic hydroxy compound  
10 is used and more preferably methanol or ethanol.

In general the ratio between the active  
hydrogen-containing compound R-H and 1,3-butadiene  
(hereinafter referred to as "reactants ratio" and  
15 expressed in moles of R-H per mole of 1,3-butadiene) is  
higher than 0.5. According to the present invention the  
active hydrogen-containing compound R-H is used in an  
amount of preferably 1 to 20 moles, and more preferably  
20 1 to 10 moles per mole of 1,3-butadiene.

The telomerization catalyst to be used in step  
(i) of the present invention can be any catalyst which,  
when contacting the reactants in its presence, promotes  
the formation of a 1-substituted-2,7-octadiene.  
25 Examples of telomerization catalysts are the transition  
metals Fe, Co, Ni, Ru, Rh, Pd, Os, Ir and Pt  
(hereinafter referred to as "Group VIII transition  
metals") and compounds thereof, including those  
30 supported on an inert carrier. Suitable examples of  
such catalyst compounds for use in step (i) of the  
process of the present invention are described in an  
article of A. Behr titled "Homogeneous Transition Metal  
Catalysts" in Aspects of Homogeneous Catalysis (1984),  
Vol. 5, pages 3-73, as well as in the references cited



hereinbefore. It is preferred that the catalyst is homogeneous under reaction conditions.

Preferably a nickel, rhodium, palladium or platinum catalyst is used as the Group VIII transition metal catalyst, more preferably a palladium catalyst. Both palladium(II) compounds and palladium(0) complexes may be used in the present process. Examples of suitable palladium(II) compounds include: palladium(II) salts, such as palladium bromide, palladium chloride, palladium formate, palladium acetate, palladium propionate, palladium octanoate, palladium carbonate, palladium sulfate, palladium nitrate, palladium borate, palladium citrate, palladium hydroxide, palladium aryl- or alkylsulfonates, sodium or potassium chloropalladate; compounds of formula  $PdX_2L_2$  in which X is a monovalent acid residue and L is a ligand selected from the group of organic nitriles, such as benzonitrile, tertiary amines, e.g. pyridine, quinoline, picoline, lutidine and triethylamine, an example of such a  $PdX_2L_2$  compound being dichlorobis(benzonitrile) palladium; palladium chelates such as palladium acetylacetonate; n-allyl complexes, like n-allyl palladium chloride or acetate, and bis (n-allyl) palladium; and 1,5-cyclooctadiene palladium chloride. It is noted that in case a palladium halide is used, also a catalyst activator is required, in order to dissociate the palladium halide in the solution so as to form a catalytically active species. Suitable palladium(0) complexes include a wide variety of substances containing as ligands, for example phosphines, alkenes, dienes, or nitriles. Examples of these include tetrakis(triphenylphosphine) palladium, bis(1,5-cyclooctadiene) palladium, and the like.

It is advantageous to employ in step (i) of the present process, besides the Group VIII transition metal catalyst, a tertiary phosphorous, arsenic or antimony ligand compound. Suitable ligand compounds comprise the substituted hydrides of phosphorous, arsenic or antimony and especially those in which the hydride is fully substituted by alkyl, alkoxy, cycloalkyl, cycloalkoxy, aryl and/or aryloxy groups. Also ligand compounds in which two phosphorous, arsenic or antimony atoms, respectively, are attached to each other through an alkylene group can be used. Examples of the latter compounds include diphosphines, diarsines and distibines. It is preferred in step (i) of the present invention to use a catalyst system comprising a palladium catalyst and a tertiary phosphorous ligand compound.

Tertiary phosphines which may be used advantageously include phosphines containing (cyclo)alkyl and/or aryl groups. Examples of suitable trialkylphosphines are triethylphosphine, tri-n-butylphosphine, tri-isobutylphosphine, tripentylphosphine, trihexylphosphine, trioctylphosphine, trilaurylphosphine, lauryldimethylphosphine, hexyldipropylphosphine and ethylbutyloctylphosphine. Examples of suitable phosphines containing one or more aromatic groups or cycloalkyl groups are tricyclohexylphosphine, triphenylphosphine, tri(p-tolyl)phosphine, dimethylphenylphosphine, methyldiphenylphosphine, tri(m-ethylphenyl)phosphine and tri(2,4-dimethylphenyl)phosphine. A suitable diphosphine which may be used is bis(diphenylphosphine)-ethane. Also the arsine and antimony homologs may be advantageously used. The above-mentioned ligand

compounds may in addition be substituted by halogen atoms, such as chlorine atoms, nitrile and/or nitro groups and the like. Also ligand compounds which contain in addition hydrophilic groups, such as -COOM, -SO<sub>3</sub>M, and/or -NH<sub>2</sub> groups attached directly or via an alkylene group to the phosphine substituent(s), may be used. In the above-mentioned groups M represents an inorganic or organic cationic residue, such as an alkali or alkaline earth metal cation, a quaternary ammonium ion or the like.

It has been found advantageous to carry out the telomerization reaction in the presence of a basic compound (hereinafter referred to as "catalyst promoter"). Suitable catalyst promoters include alkali metal, alkaline earth metal or quaternary ammonium hydroxides, alcoholates or alkoxides, enolates, phenoxides, alkali metal borohydride, hydrazines and the like. As catalyst promoter preferably an alkali metal salt, especially the sodium or potassium salt of the active hydrogen-containing compound R-H is used. These alkali metal salts, for example, may be added as such or may be formed in situ by dissolving an alkali metal in the active hydrogen-containing compound.

The amount of Group VIII transition metal catalyst to be used is not critical, and any catalytically effective amount may be used. In general, amounts between 0.000001 and 1, and preferably between 0.000005 and 0.01 gramatom of Group VIII transition metal catalyst per mole of 1,3-butadiene can be used. In order to achieve high catalyst efficiencies without having to recycle the catalyst, amounts of less than 0.0001, preferably less than 0.00005, and more preferably less than 0.00002 gramatoms of Group VIII

transition metal catalyst are used per mole of 1,3-butadiene.

The ligand compound is generally used in a relative amount of 1 to 20 moles, and preferably 2 to 15 moles of ligand compound per gramatom of the Group VIII transition metal of the catalyst. The ligand compound may be added as a separate compound to the reaction mixture or zone or to the catalyst make-up solution, or it may be incorporated in a Group VIII transition metal catalyst complex. When, for example, tetrakis(triphenylphosphine) palladium is used as catalyst compound, it is generally not necessary to add additional triphenylphosphine as ligand compound. The catalyst promoter is usually employed in a relative amount of 0.5 to 1,000 moles, and more preferably 0.5-150 moles of promoter per gramatom of the Group VIII transition metal catalyst.

The process according to the present invention should preferably be conducted in the substantial absence of oxygen, as oxygen reacts with the ligand compound and consequently may result in decreased catalyst activity.

Step (i) of the process of the present invention is conducted in the liquid phase and may be carried out in the presence or absence of a solvent which is inert under reaction conditions. When one or more of the reactants are not liquid under reaction conditions or if the reactants are poorly miscible or substantially immiscible with each other then preferably a solvent is used to secure the desired homogeneous system. Any solvent that will solubilize 1,3-butadiene, the active hydrogen-containing compound and the

catalyst, ligand and optional promoter components may be used in the present process. Suitable inert solvents are (cyclo)alkanes, for example C<sub>6-12</sub>-alkanes and cyclohexane or a mixture thereof; aromatic compounds, e.g. benzene, toluene, xylene, naphthenes; a polar solvent, such as tertiary alcohols, an amide, e.g. acetamide, dimethylacetamide or dimethylformamide; a nitrile compound, such as acetonitrile or benzonitrile; a ketone, e.g. acetone or methyl ethylketone; an ester compound, such as ethylene glycoldiacetate; an ether compound, e.g. ethylether, propylether, dioxane, tetrahydrofuran, and anisole, or a lower alkyl-disubstituted ether of mono- or diethylene glycol; and dimethylsulfoxide and sulpholane. Also water may be used as solvent or co-solvent.

The temperature at which the telomerization reaction is carried out is not critical. Normally, temperatures between ambient temperature and 150°C can be used. Preferably, the reaction temperature is 50-100°C and more preferably 70-100°C.

The pressure at which the telomerization reaction is carried out is not critical. Generally the reaction pressure lies between 1 and 40 bars, preferably between 5 and 30 bars and most preferably between 10 and 20 bars.

In general the reaction time is not critical. Reaction times of from 0.1 to 100 hours give good results, depending on reaction temperatures, reactants, catalyst system and optional solvent. Preferably reaction times lie between 0.1 and 20 hours and more preferably between 0.2 and 10 hours.

In process step (i) any type of reactor may be used. Examples of suitable reactors are a stirred tank reactor and a tubular reactor.

Step (i) of the process of the present invention generally produces a mixture of 1-substituted-2,7-octadiene, 3-substituted-1,7-octadiene and 1,3,7-octatriene, as well as a minor amount of high boiling components. Substantially no branched C<sub>8</sub>-products are formed.

It is advantageous to recover the 1-substituted-2,7-octadiene from the reaction mixture by means of known techniques. Depending on the physical properties of reactants, catalyst components, optional solvent and the reaction products, the desired product (1-substituted-2,7-octadiene) may be isolated by distillation, extraction or other well known separation techniques, or a combination thereof. Preferably, the reaction mixture is subjected to one or more distillation steps in order to recover the 1-substituted-2,7-octadiene. The number and nature of the distillation steps will depend on the relative boiling points of the 1- and 3-substituted octadienes, the octatriene, the active hydrogen-containing compound R-H and the optional solvent. The 1,3-butadiene, and other components present in a crude C<sub>4</sub>-hydrocarbon mixture, if used instead of pure 1,3-butadiene, have a significantly lower boiling point than the reaction products and thus can be easily separated. On the other hand, the high boiling compounds and the catalyst or catalyst decomposition products generally remain in the distillation residue.

At this point in the process the 3-substituted-1,7-octadiene and/or the 1,3,7-octatriene can advantageously also be separated from the 1-substituted-2,7-octadiene, such as for example by well known distillation techniques. It may, however, be more  
5 convenient to separate the 3-substituted-1,7-octadiene and/or 1,3,7-octatriene after they have been subjected to step (ii) of the present process, the hydrogenation step. This could be the case, for example, if the  
10 difference in the physical property relevant for the separation technique, e.g. in boiling point, between the 3-substituted octane and/or octane, which are the hydrogenation products of the 3-substituted-1,7-octadiene and 1,3,7-octatriene, and the 1-substituted  
15 octane, which is the hydrogenation product of the 1-substituted-2,7-octadiene, would be greater than the difference in the physical property relevant for the separation technique, e.g. in boiling point, between the 3-substituted-2,7-octadiene and/or 1,3,7-octatriene and  
20 the 1-substituted-1,7-octadiene.

The 1,3-butadiene, active hydrogen-containing compound R-H and optional solvent recovered can be recycled to step (i) of the process. The catalyst can  
25 be separated from the other components in the reaction mixture by means of extraction, precipitation or distillation, in which latter case it remains in the distillation residue. The catalyst in the residue or  
30 extractant may subsequently be regenerated, if desired, and recycled to step (i) or be separated from said residue or extractant, respectively, by means of precipitation, filtration, extraction or combination thereof. These catalyst recovery and recycling techniques are all well known in the art.

The 1-substituted-2,7-octadiene formed in process step (i), optionally containing minor amount of the by-products 3-substituted-1,7-octadiene and/or 1,3,7-octatriene, is subjected to hydrogenation in process step (ii) to form the 1-substituted octane of formula  $\text{CH}_3\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-R}$ , optionally together with the 3-substituted octane and/or n-octane.

In general, any conventional hydrogenation process can be used. The hydrogenation may be carried out in the liquid phase or in the vapor phase. Depending on the nature of the starting material, the reaction can be carried out at a temperature between 0 and 400°C. Preferably the temperature is in the range from ambient to 350°C. More preferably the hydrogenation is carried out at a temperature between 50 and 200°C. The pressure is not critical and depends on whether the hydrogenation is carried out in the liquid or vapor phase. In general the pressure can vary between 0.1 and 100 bar.

In process step (ii) any of the conventional homogeneous and heterogeneous hydrogenation catalysts can be used. Examples of such catalysts include the Group VIII of the periodic system transition metal catalysts. Preferred catalysts are the noble metal catalysts of this group VIII, either in elemental form or in the form of a compound which is reducible by hydrogen under hydrogenation conditions to form finely divided elemental metal, or mixtures thereof. Examples of such compounds are the oxides, halides, nitrates, oxalates, acetates, carbamates, propionates, tartrates, hydroxides and the like or mixtures thereof. Preferred catalysts are palladium and platinum, whereas also Raney-nickel may advantageously be used. These



catalysts may be modified with other metals. The catalysts may preferably be supported on a carrier material, such as for example active carbon, alumina, silica, silica-alumina, calcium carbonate, barium carbonate, barium sulfate, clays, Kieselguhr and the like. Particularly preferred catalysts are alumina supported platinum or palladium catalysts, the metal preferably being dispersed on the support in an amount of 0.1 to 1.0 wt%.

10       The hydrogenation catalysts are used in the amounts conventional for these type of reactions and catalysts.

15       Generally, the relative amounts of hydrogen, in the form of pure hydrogen or as a hydrogen containing gas, and the 1-substituted-2,7-octadiene are not critical, but it is preferred to use a molar ratio between hydrogen and the 1-substituted-2,7-octadiene in the range from 2:1 to 50:1, more preferably from 2:1 to 20       10:1.

25       The hydrogenation can advantageously be carried out in a diluent or solvent for dilution of the material to be hydrogenated. The usual diluents may be used for this, such as saturated hydrocarbons having 4 to 12 carbon atoms, alkanols containing from 1 to 12 carbon atoms, acyclic and cyclic ethers having 4 to 12 carbon atoms and mixtures thereof. Advantageously, part of the 30       reaction product of the hydrogenation reaction is recycled to step (ii) and used as diluent in the hydrogenation reaction.

      The hydrogenation in step (ii) can be carried out according to batch, semi-continuous or continuous

procedures. It may also be performed in more than one step, such as for example in a preliminary hydrogenation step and a finishing hydrogenation step. Typical weight hourly space velocities (based on reactant) are in the range of 0.1-50, preferably in the range of 0.2-10 and  
5 more preferably in the range of 0.5-5.

The 1-substituted-2,7-octadiene formed in process step (i), optionally containing 3-substituted-1,7-octadiene and/or 1,3,7-octatriene is hydrogenated  
10 in step (ii) of the present process at very high conversion and selectivity. Apart from the reaction product 1-substituted octane and, depending on the feed composition, 3-substituted octane and n-octane, minor  
15 amounts of 1-octene and its internal linear isomers may be formed due to decomposition of the 1- or 3-substituted octanes under hydrogenation conditions. Accordingly, both the telomerization and hydrogenation  
products obtained by the process of the present  
20 invention contain substantially no branched C<sub>8</sub>-isomers.

Depending on the physical properties of the starting compound(s), catalyst, optional diluent and the reaction product(s), the desired product (1-substituted  
25 octane) may be isolated by distillation, extraction or other well known separation techniques, or a combination thereof. The reaction mixture may advantageously be subjected to one or more distillation steps in order to recover the 1-substituted octane. The number and nature  
30 of the distillation steps will depend on the relative boiling points of the 1- and 3-substituted octanes, the n-octane and the optional solvent.

It is a matter of convenience and it will depend on the physical properties of the components

present in the hydrogenation reaction mixture, whether the desired reaction product (1-substituted octane) is recovered in a substantially pure form (>99%) and as such used in step (iii) of the present process, or alternatively, a mixture of reaction products is used in the last process step. The latter would be preferred, for example, if the difference in the physical property relevant for the separation technique, e.g. in boiling point, between the 3-substituted octane on the one hand, and the 1-substituted octane, on the other hand, would be smaller than the difference in the relevant physical property, e.g. in boiling point, between the decomposition (in step (iii)) products of 3-substituted octane, on the one hand, and 1-octene, on the other hand. At the high conversions and selectivities, apart from the removal of unconsumed hydrogen, and optional solvent, no further purification of the 1-substituted octane is necessary if the 1-substituted-2,7-octadiene is subjected to the hydrogenation step in a pure form ( $\geq 99\%$ )

Generally, it is preferred to purify the 1-substituted octane as much as possible, so that the amount of 3-octene, which is one of the decomposition products of the 3-substituted octane, is minimal; as 3-octene will be relatively difficult to separate from 1-octene.

The catalyst can be separated (if not present in a fixed bed) from the reaction mixture by extraction, distillation or conventional filtration means. The reactants and/or diluent can be recycled to the hydrogenation reaction, if desired. As mentioned

before, it is advantageous to recycle the reaction product in order to serve as diluent.

It is noted that the group R in the 1-substituted-2,7-octadiene compound to be hydrogenated can also be a group which can be hydrogenated under the prevailing hydrogenation conditions, provided the thus hydrogenated group R is readily split off in the last step of the process to yield 1-octene.

According to process step (iii) of the present invention, the 1-substituted octane of formula  $\text{CH}_3\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-R}$  formed in step (ii) is subsequently decomposed in the presence of a suitable catalyst to form 1-octene.

The type of decomposition reaction which is the subject of process step (iii) is in itself well known in the art. In process step (iii) of the present invention any suitable conventional catalyst can be used which is capable of decomposing the 1-substituted octane to give 1-octene. Generally, a solid acidic catalyst is used for this purpose. Preferably an alumina catalyst is used. Examples of such catalysts are alpha, delta, gamma, etha and theta-aluminas, which may be modified with metal-containing weakly acidic components, organic carboxylic acids, or other treating agents.

The temperatures at which the decomposition process is carried out depend on the decomposition temperature of the respective compound and on the catalyst activity. Generally the temperatures are up to 500°C, preferably in the range of 200 to 400°C, whereas temperatures in the range of 250 to 350°C are more preferred.

The pressure under which the decomposition reaction may be carried out is not critical and can vary widely, for example from 0.1 to about 50 bar. Preferably the reaction is carried out at a pressure between 0.2 and 10 bar.

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The reaction can be carried out in the vapor or in the liquid phase, the vapor phase being preferred. An inert carrier gas or an inert diluent may be used to dilute the reactant. Examples of such inert gases are  
10 nitrogen, helium, argon and the like or mixtures thereof.

The reaction may be carried out continuously, semi-continuously or batchwise. In the continuous mode  
15 the reactant(s) and possibly the diluent are continuously passed over a bed of the catalyst at the desired reaction conditions. The reactant is generally added to the reactor containing the catalyst under the desired reaction conditions at a weight hourly space  
20 velocity of about 0.01 to about 50, and preferably about 0.1 to about 10.

The desired product 1-octene may be recovered from the reaction mixture using conventional methods  
25 such as solvent extraction, (fractional) distillation, (fractional) condensation etc. The unconverted reactant(s) can be recycled to this reaction step (iii). The other product of the decomposition reaction, the compound R-H can be recovered and recycled to process  
30 step (i).

Apart from 1-octene, the 1-substituted octane may also yield some 2-octene caused by isomerization. The 3-substituted octane will be decomposed to 2-octene and 3-octene, whereas any n-octane present in the feed will be inert. Minor amounts of 4-octenes may be formed

due to isomerization of 3-octenes. Accordingly, the decomposition product mixture contains 1-octene and minor quantities of 2-, 3- and 4-octenes and n-octane; no branched C<sub>8</sub>-olefin isomers will be formed.

5 By means of well known distillation techniques the C<sub>8</sub>-olefinic fraction, usually containing some n-octane, unconverted 1-substituted octane and optionally unconverted 3-substituted octane can be recovered. This C<sub>8</sub>-olefinic fraction may be subjected to a further  
10 distillation step to remove unconverted 1- and 3-substituted octanes. This already can yield a 1-octene product having a purity of about 90-97%. Finally, in a further fractional distillation step the linear internal octene isomers and n-octane can be removed so as to give  
15 1-octene in a highly pure form (>97%). The 1-octene thus obtained contains no branched C<sub>8</sub>-olefin isomers. The typical conditions for the separation steps, and in particular the distillation steps are largely determined  
20 by the physical properties, such as in particular the boiling points of the compounds to be separated.

The invention will be illustrated by means of the following example which by no means should be construed to limit the invention.

25 Example:

A. To a tubular reactor provided with a jacket for temperature control and maintained at a pressure of 14 bar and at a temperature of 80°C, 1.3 kg/h of  
30 deoxygenated methanol, 0.8 kg/h of 1,3-butadiene, 45 g/h of a catalyst solution prepared by dissolving 18 g of palladium acetylacetonate (59 mmole) and 31 g of triphenylphosphine (118 mmole) in 16 l of deoxygenated methanol, and 30 g/h of a promoter solution containing 1.5 wt% of sodium methoxide in de oxygenated methanol

were fed. In the feed stream to the reactor the methanol/1,3-butadiene ratio was 3 mole/mole, the palladium concentration was 0.000015 gram atoms per mole of 1,3-butadiene, the triphenylphosphine/palladium ratio was 2 mole/gram atom, and the sodium methoxide/palladium ratio was 100 mole/gram atom. The average residence time in the tubular reactor was 3.5 h. Gas chromatographic analysis of the reaction mixture leaving the reactor showed that 95.7% of the 1,3-butadiene fed was converted, whereas the selectivity of the converted 1,3-butadiene to (substituted) C<sub>8</sub> products was: 89.4% 1-methoxy-2,7-octadiene, 5.6% 3-methoxy-1,7-octadiene, and 5.0% 1,3,7-octatriene.

The reaction mixture thus obtained was subjected to a first distillation at 1.2 bar to remove the lower boiling components 3-methoxy-1,7-octadiene, 1,3,7-octatriene, unreacted methanol and 1,3-butadiene. The bottom stream of the first distillation was subjected to a further distillation at 0.5 bar to remove 1-methoxy-2,7-octadiene from the heavier boiling components to obtain 1-methoxy-2,7-octadiene in a purity of 99.3%.

B. The 1-methoxy-2,7-octadiene having a purity of 99.3% obtained in process step (i) was fed to a tubular reactor containing 67 g of catalyst in a fixed bed at a rate of 60 g (0.43 mole) per hour (weight hourly space velocity = 0.9). The catalyst used was a palladium on alumina catalyst, containing 0.3 wt% palladium supported on  $\gamma$ -alumina. Hydrogen was fed to the reactor at such a rate as to maintain a molar ratio of hydrogen to 1-methoxy-2,7-octadiene of about 7:1. The temperature of the reactor was maintained at 80°C and the pressure at 15 bar to maintain a liquid phase. At the end of the reactor analysis of the reaction mixture by gas

chromatography showed that the purity of the product 1-methoxyoctane in the liquid phase was 99.2% as determined by gas chromatography.

C. The 1-methoxyoctane having a purity of 99.2% obtained in process step (ii) was subjected to a decomposition reaction. The 1-methoxyoctane was fed to a preheater at a rate of 80 g (0.55 mole) per hour in order to vaporize the 1-methoxyoctane. Subsequently, the vaporized 1-methoxyoctane was fed to a tubular reactor containing 13 g of a  $\gamma$ -alumina catalyst in a fixed bed (weight hourly space velocity = 6.2). The temperature of the reactor was maintained at 330°C and the pressure at 1 bar absolute. At the end of the reactor, analysis of the product by gas chromatography showed that the conversion of 1-methoxyoctane was 80% and the selectivity to octenes was 66% and the selectivity to 1-octanol and dioctylether was 34%. Based on octenes present in the reaction mixture, there was 95.7% of 1-octene. The reaction mixture was subjected to an atmospheric distillation to remove methanol and other components of boiling points lower than the octenes. In a second distillation (at 0.5 bar pressure) the unconverted 1-methoxyoctane, 1-octanol and dioctylether were removed to yield an octene fraction consisting of 95.7 wt% 1-octene, 3.8 wt% 2-octenes and 0.5 wt% 3- and 4-octenes. This octene fraction was subjected to a final distillation step at a pressure of 0.5 bar to give an octene fraction of 99.0 wt% 1-octene, 0.55 wt% 2-octenes and 0.45 wt% 3- and 4-octenes.



## PATENT CLAIMS:

1. Process for producing 1-octene which comprises the steps of:

- 5 (i) reacting 1,3-butadiene with a compound containing an active hydrogen atom and having formula R-H, in the presence of a telomerization catalyst to form a 1-substituted-2,7-octadiene of formula  $\text{CH}_2=\text{CH}-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}=\text{CH}-\text{CH}_2-\text{R}$ , in which R represents the residue of the compound containing an active hydrogen atom;
- 10 (ii) subjecting the 1-substituted-2,7-octadiene formed in step (i) to hydrogenation in the presence of a hydrogenation catalyst to form a 1-substituted octane of formula  $\text{CH}_3-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{R}$ ; and
- 15 (iii) decomposing the 1-substituted octane formed in step (ii) in the presence of a suitable catalyst to form 1-octene.

2. Process according to Claim 1, wherein in step (i) a primary aliphatic alcohol or aromatic hydroxy compound is used as the active hydrogen-containing compound R-H.

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3. Process according to Claim 2, wherein in step (i) as primary aliphatic alcohol methanol or ethanol is used.

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4. Process according to any one of the Claims 1-3, wherein in step (i) a telomerization catalyst comprising palladium and a tertiary phosphorous ligand compound is used.
- 5 5. Process according to any one of the Claims 1-4, wherein in step (i) the reaction is carried out in the presence of a catalyst promotor.
- 10 6. Process according to Claim 5, wherein in step (i) as catalyst promotor an alkali metal salt of the active hydrogen-containing compound R-H is used.
- 15 7. Process according to any one of the Claims 1-6, wherein in step (ii) the hydrogenation is carried out in a diluent at a temperature between 0 and 400°C in the presence of a hydrogenation catalyst, and the molar ratio of hydrogen to 1-substituted-2,7-octadiene is in the range from 2:1 to 50:1.
- 20 8. Process according to Claim 7, wherein in step (ii) the hydrogenation catalyst is a supported platinum or palladium catalyst.
- 25 9. Process according to Claims 7 or 8, wherein part of the reaction product of the hydrogenation reaction is recycled to step (ii) and used as diluent in the hydrogenation reaction.
- 30 10. Process according to any one of the Claims 1-9, wherein in step (iii) the decomposition reaction is carried out in the vapor phase at a temperature up to 500°C over a fixed bed of an alumina catalyst.

## AMENDED CLAIMS

[received by the International Bureau on 23 September 1991 (23.09.91); original claim 2 deleted; original claim 1 amended; remaining claims unchanged but renumbered as claims 2-9 (2 pages)]

1. Process for producing 1-octene which comprises the steps of:

(i) reacting 1,3-butadiene with a primary aliphatic alcohol or aromatic hydroxy compound having formula R-H, in the presence of a telomerization catalyst to form a 1-substituted-2,7-octadiene of formula  $\text{CH}_2=\text{CH}-\text{CH}_2-\text{CH}_2-\text{CH}=\text{CH}-\text{CH}_2-\text{R}$ , in which R represents the residue of the primary aliphatic alcohol or aromatic hydroxy compound;

(ii) subjecting the 1-substituted-2,7-octadiene formed in step (i) to hydrogenation in the presence of a hydrogenation catalyst to form a 1-substituted octane of formula  $\text{CH}_3-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{R}$ ; and

(iii) decomposing the 1-substituted octane formed in step (ii) in the presence of a suitable catalyst to form 1-octene.

2. Process according to Claim 1, wherein in step (i) as primary aliphatic alcohol methanol or ethanol is used.

## 26

3. Process according to any one of the Claims 1-2, wherein in step (i) a telomerization catalyst comprising palladium and a tertiary phosphorous ligand compound is used.

4. Process according to any one of the Claims 1-3, wherein in step (i) the reaction is carried out in the presence of a catalyst promotor.

5. Process according to Claim 4, wherein in step (i) as catalyst promotor an alkali metal salt of the active hydrogen-containing compound R-H is used.

6. Process according to any one of the Claims 1-5, wherein in step (ii) the hydrogenation is carried out in a diluent at a temperature between 0 and 400°C in the presence of a hydrogenation catalyst, and the molar ratio of hydrogen to 1-substituted-2,7-octadiene is in the range from 2:1 to 50:1.

7. Process according to Claim 6, wherein in step (ii) the hydrogenation catalyst is a supported platinum or palladium catalyst.

8. Process according to Claims 6 or 7, wherein part of the reaction product of the hydrogenation reaction is recycled to step (ii) and used as diluent in the hydrogenation reaction.

9. Process according to any one of the Claims 1-8, wherein in step (iii) the decomposition reaction is carried out in the vapor phase at a temperature up to 500°C over a fixed bed of an alumina catalyst.

# INTERNATIONAL SEARCH REPORT

International Application No **PCT/EP 90/02194**

<b>I. CLASSIFICATION OF SUBJECT MATTER</b> (If several classification symbols apply, indicate all) * According to International Patent Classification (IPC) or to both National Classification and IPC IPC <sup>5</sup> : <b>C 07 C 11/02</b>								
<b>II. FIELDS SEARCHED</b> <div style="text-align: center; border-top: 1px solid black; border-bottom: 1px solid black;">Minimum Documentation Searched <sup>7</sup></div> <table style="width: 100%; border-collapse: collapse;"> <tr> <td style="width: 25%; border-bottom: 1px solid black;">Classification System</td> <td style="border-bottom: 1px solid black;">Classification Symbols</td> </tr> <tr> <td style="padding: 5px;">IPC<sup>5</sup></td> <td style="padding: 5px; text-align: center;">C 07 C</td> </tr> </table> <div style="border-top: 1px solid black; padding-top: 5px;">Documentation Searched other than Minimum Documentation to the extent that such Documents are included in the Fields Searched <sup>8</sup></div>			Classification System	Classification Symbols	IPC <sup>5</sup>	C 07 C		
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IPC <sup>5</sup>	C 07 C							
<b>III. DOCUMENTS CONSIDERED TO BE RELEVANT <sup>9</sup></b> <table style="width: 100%; border-collapse: collapse;"> <tr> <th style="width: 10%; border-bottom: 1px solid black;">Category <sup>9</sup></th> <th style="width: 60%; border-bottom: 1px solid black;">Citation of Document, <sup>11</sup> with indication, where appropriate, of the relevant passages <sup>12</sup></th> <th style="width: 30%; border-bottom: 1px solid black;">Relevant to Claim No. <sup>13</sup></th> </tr> <tr> <td style="vertical-align: top; padding: 5px; text-align: center;">X</td> <td style="padding: 5px;">           Chemical Abstracts, volume 114, 7 January 1991, (Columbus, Ohio, US),            see page 574, abstract no. 5796w            &amp; JP, A, 02172924 (KURARAY CO. LTD.)            4 July 1990   <div style="text-align: center;">-----</div> </td> <td style="vertical-align: top; padding: 5px; text-align: center;">1,8</td> </tr> </table>			Category <sup>9</sup>	Citation of Document, <sup>11</sup> with indication, where appropriate, of the relevant passages <sup>12</sup>	Relevant to Claim No. <sup>13</sup>	X	Chemical Abstracts, volume 114, 7 January 1991, (Columbus, Ohio, US), see page 574, abstract no. 5796w & JP, A, 02172924 (KURARAY CO. LTD.) 4 July 1990  <div style="text-align: center;">-----</div>	1,8
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<div style="display: flex; justify-content: space-between;"> <div style="width: 45%;"> <p>* Special categories of cited documents: <sup>10</sup></p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> </div> <div style="width: 50%;"> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.</p> <p>"A" document member of the same patent family</p> </div> </div>								
<b>IV. CERTIFICATION</b> <table style="width: 100%; border-collapse: collapse;"> <tr> <td style="width: 50%; border-bottom: 1px solid black; padding: 5px;">           Date of the Actual Completion of the International Search  <div style="text-align: center;">31st July 1991</div> </td> <td style="width: 50%; border-bottom: 1px solid black; padding: 5px;">           Date of Mailing of this International Search Report  <div style="text-align: center;">30. 08. 91</div> </td> </tr> <tr> <td style="border-bottom: 1px solid black; padding: 5px;">           International Searching Authority  <div style="text-align: center;">EUROPEAN PATENT OFFICE</div> </td> <td style="border-bottom: 1px solid black; padding: 5px;">           Signature of Authorized Officer  <div style="display: flex; align-items: center;"> <div style="border: 1px solid black; padding: 2px 5px; margin-right: 10px;">M. PEIS</div> <div style="font-family: cursive; font-size: 1.2em;">M. Peis</div> </div> </td> </tr> </table>			Date of the Actual Completion of the International Search <div style="text-align: center;">31st July 1991</div>	Date of Mailing of this International Search Report <div style="text-align: center;">30. 08. 91</div>	International Searching Authority <div style="text-align: center;">EUROPEAN PATENT OFFICE</div>	Signature of Authorized Officer <div style="display: flex; align-items: center;"> <div style="border: 1px solid black; padding: 2px 5px; margin-right: 10px;">M. PEIS</div> <div style="font-family: cursive; font-size: 1.2em;">M. Peis</div> </div>		
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